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## Structure Reports

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2-Fluoro-*N*-*o*-tolylbenzamideAamer Saeed,<sup>a\*</sup> Rasheed Ahmad Khera,<sup>a</sup> Shahid Ameen,<sup>a</sup>  
Jim Simpson<sup>b</sup> and Roderick G. Stanley<sup>b</sup><sup>a</sup>Department of Chemistry, Quaid-i-Azam University, Islamabad 45320, Pakistan, and <sup>b</sup>Department of Chemistry, University of Otago, PO Box 56, Dunedin, New Zealand

Correspondence e-mail: aamersaeed@yahoo.com

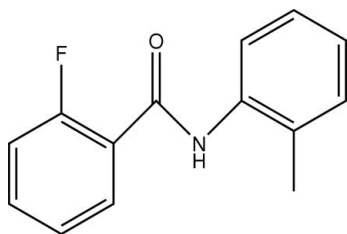
Received 15 December 2008; accepted 18 December 2008

Key indicators: single-crystal X-ray study;  $T = 93$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å; disorder in main residue;  $R$  factor = 0.037;  $wR$  factor = 0.108; data-to-parameter ratio = 7.3.

In the title compound,  $\text{C}_{14}\text{H}_{12}\text{FNO}$ , the *ortho*-F atom and corresponding H atom on the fluorobenzene ring are disordered over two positions with occupancies of 0.856 (4) and 0.144 (4). The amide unit is planar with a maximum deviation of 0.0057 (16) Å and the amide plane makes dihedral angles of 38.27 (11)° with the fluorobenzene ring plane and 37.53 (10)° with the tolyl ring. The two benzene rings are inclined at an angle of 4.17 (15)°. In the crystal structure, chains form along *b* through  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds augmented by  $\text{C}-\text{H}\cdots\pi$  interactions. Additional intermolecular  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{F}$  hydrogen bonds further stabilize the structure, forming layers in the *ac* plane.

## Related literature

For related structures, see: Chopra & Guru Row (2008); Donnelly *et al.* (2008); Hou *et al.* (2004); Saeed *et al.* (2008). For reference structural data, see: Allen *et al.* (1987).



## Experimental

## Crystal data

$\text{C}_{14}\text{H}_{12}\text{FNO}$   
 $M_r = 229.25$   
 Monoclinic,  $P2_1/n$   
 $a = 10.749$  (4) Å  
 $b = 4.8245$  (17) Å  
 $c = 21.580$  (7) Å  
 $\beta = 93.820$  (19)°

$V = 1116.6$  (7) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 93$  (2) K  
 $0.25 \times 0.15 \times 0.05$  mm

## Data collection

Bruker APEXII CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker, 2006)  
 $T_{\min} = 0.779$ ,  $T_{\max} = 0.995$

7896 measured reflections  
 1228 independent reflections  
 925 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.067$   
 $\theta_{\text{max}} = 21.3^\circ$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$   
 $wR(F^2) = 0.108$   
 $S = 1.06$   
 1228 reflections  
 169 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.19$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H}\text{N1}\cdots\text{O1}^{\text{i}}$	0.85 (3)	2.08 (3)	2.887 (3)	158 (2)
$\text{C6}-\text{H6}\cdots\text{F1}^{\text{ii}}$	0.95	2.60	3.219 (11)	123
$\text{C12}-\text{H12}\cdots\text{O1}^{\text{iii}}$	0.95	2.63	3.434 (3)	143
$\text{C12}-\text{H12}\cdots\text{F1}^{\text{iii}}$	0.95	2.40	3.287 (11)	154
$\text{C14}-\text{H14C}\cdots\text{Cg1}^{\text{i}}$	0.98	2.98	3.702 (3)	131

Symmetry codes: (i)  $x, y - 1, z$ ; (ii)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (iii)  $-x + 1, -y + 3, -z$ . Cg1 is the centroid of the C8–C13 benzene ring.

Data collection: APEX2 (Bruker 2006); cell refinement: APEX2 and SAINT (Bruker 2006); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008) and TITAN2000 (Hunter & Simpson, 1999); molecular graphics: SHELXTL (Sheldrick, 2008) and Mercury (Macrae *et al.*, 2006); software used to prepare material for publication: SHELXL97, enCIFer (Allen *et al.*, 2004), PLATON (Spek, 2003) and publCIF (Westrip, 2009).

We thank the University of Otago for purchase of the diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2882).

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**supplementary materials**

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## 2-Fluoro-*N*-*o*-tolylbenzamide

A. Saeed, R. A. Khera, S. Ameen, J. Simpson and R. G. Stanley

### Comment

As part of our ongoing work on the structure of benzanilides and related compounds (Saeed *et al.*, 2008), we report here the structure of the title 2-fluorobenzamide derivative, (I), Fig 1. The C2—C1—O1—N1—C8 unit is almost planar with a maximum deviation of 0.0057 (16) Å for N1. This plane makes dihedral angles of 38.27 (11) ° with the fluorobenzene ring plane and 37.53 (10)° with the tolyl ring. The two benzene rings are inclined at an angle of 4.17 (15)° giving the molecule a stepped structure. Bond distances in the molecule are not unusual (Allen *et al.*, 1987) and agree well with those reported previously (see for example Chopra & Guru Row, 2008; Donnelly *et al.*, 2008; Hou *et al.*, 2004, Saeed *et al.*, 2008).

In the crystal structure, chains form along *b* through N—H···O hydrogen bonds augmented by C—H··· $\pi$  interactions involving the methyl group and the tolyl rings (Table 1, Fig. 2). Additional intermolecular C—H···O and C—H···F hydrogen bonds further stabilize the structure forming layers in the *ac* plane, that stack down the *b* axis (Fig. 3).

### Experimental

2-Fluorobenzoyl chloride (5.4 mmol) in CHCl<sub>3</sub> was treated with 2-methylaniline (21.6 mmol) under a nitrogen atmosphere at reflux for 3 h. Upon cooling, the reaction mixture was diluted with CHCl<sub>3</sub> and washed consecutively with 1 *M* aqueous HCl and saturated aq NaHCO<sub>3</sub>. The organic layer was dried over anhydrous magnesium sulfate and concentrated under reduced pressure. Crystallization of the residue from CHCl<sub>3</sub> afforded (I) as colourless needles in an 81% yield: Anal. calcd. for C<sub>14</sub>H<sub>12</sub>FNO: C 73.35, H 5.28, N 6.11%; found: C 73.30, H 5.32, N 6.09%

### Refinement

The H atom bound to N1 was located in a difference electron density map and refined freely with an isotropic displacement parameter. All other H-atoms were refined using a riding model with C—H = 0.95 Å,  $U_{\text{iso}} = 1.2U_{\text{eq}}$  (C) for aromatic and 0.98 Å,  $U_{\text{iso}} = 1.5U_{\text{eq}}$  (C) for methyl H atoms. The *ortho*-F atom and corresponding H atom on the fluorobenzene ring are disordered over two positions with occupancies 0.856 (4) and 0.144 (4). Crystals were very small and weakly diffracting, with no significant data obtained beyond  $\theta = 21^\circ$

### Figures

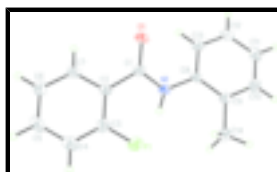


Fig. 1. The molecular structure of (I) with displacement ellipsoids for the non-hydrogen atoms drawn at the 50% probability level. For clarity atoms of the minor disorder component have been omitted.

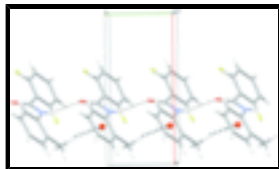


Fig. 2. Chains formed along *a* by N—H···O hydrogen bonds (dashed lines) and C—H··· $\pi$  interactions (dotted lines). Spheres represent the centroids of C8···C13 benzene rings.

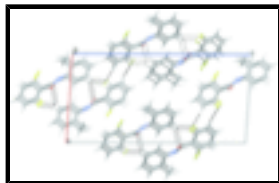


Fig. 3. Crystal packing of (I) viewed down the *b* axis with hydrogen bonds drawn as dashed lines.

## 2-Fluoro-*N*-*o*-tolylbenzamide

### Crystal data

$C_{14}H_{12}FNO$

$M_r = 229.25$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P\ 2_1n$

$a = 10.749\ (4)\ \text{\AA}$

$b = 4.8245\ (17)\ \text{\AA}$

$c = 21.580\ (7)\ \text{\AA}$

$\beta = 93.820\ (19)^\circ$

$V = 1116.6\ (7)\ \text{\AA}^3$

$Z = 4$

$F_{000} = 480$

$D_x = 1.364\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation

$\lambda = 0.71073\ \text{\AA}$

Cell parameters from 1374 reflections

$\theta = 3.3\text{--}21.2^\circ$

$\mu = 0.10\ \text{mm}^{-1}$

$T = 93\ (2)\ \text{K}$

Rectangular plate, colourless

$0.25 \times 0.15 \times 0.05\ \text{mm}$

### Data collection

Bruker APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 93\ (2)\ \text{K}$

$\omega$  scans

Absorption correction: Multi-scan  
(SADABS; Bruker, 2006)

$T_{\min} = 0.779$ ,  $T_{\max} = 0.995$

7896 measured reflections

1228 independent reflections

925 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.067$

$\theta_{\max} = 21.3^\circ$

$\theta_{\min} = 1.9^\circ$

$h = -10 \rightarrow 10$

$k = -4 \rightarrow 4$

$l = -22 \rightarrow 22$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.037$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$wR(F^2) = 0.108$	$w = 1/[\sigma^2(F_o^2) + (0.0633P)^2]$
$S = 1.06$	where $P = (F_o^2 + 2F_c^2)/3$
1228 reflections	$(\Delta/\sigma)_{\max} < 0.001$
169 parameters	$\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\min} = -0.21 \text{ e } \text{\AA}^{-3}$
	Extinction correction: none

### Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.5852 (2)	1.1330 (5)	0.13062 (12)	0.0203 (7)	
O1	0.57188 (16)	1.3797 (3)	0.11721 (8)	0.0291 (6)	
C2	0.5376 (2)	1.0250 (4)	0.18985 (11)	0.0177 (7)	
C7	0.4271 (2)	1.1328 (5)	0.20975 (13)	0.0229 (7)	
H7	0.3828	1.2677	0.1850	0.027*	0.856 (4)
F1	0.71053 (15)	0.7317 (3)	0.21278 (8)	0.0296 (7)	0.856 (4)
C3	0.5996 (2)	0.8335 (5)	0.22854 (13)	0.0242 (7)	
H3	0.6757	0.7577	0.2162	0.029*	0.144 (4)
F1'	0.3678 (10)	1.321 (2)	0.1771 (6)	0.046 (4)	0.144 (4)
C4	0.5566 (3)	0.7481 (5)	0.28378 (13)	0.0302 (7)	
H4	0.6022	0.6179	0.3093	0.036*	
C5	0.4450 (3)	0.8563 (5)	0.30152 (13)	0.0284 (7)	
H5	0.4126	0.7979	0.3393	0.034*	
C6	0.3806 (2)	1.0486 (5)	0.26455 (12)	0.0270 (7)	
H6	0.3043	1.1228	0.2769	0.032*	
N1	0.64105 (18)	0.9476 (5)	0.09499 (10)	0.0197 (6)	
HN1	0.639 (2)	0.781 (5)	0.1077 (12)	0.028 (8)*	
C8	0.6910 (2)	1.0053 (4)	0.03731 (11)	0.0173 (7)	
C9	0.8010 (2)	0.8696 (4)	0.02298 (12)	0.0186 (7)	
C14	0.8660 (2)	0.6685 (5)	0.06812 (12)	0.0243 (7)	
H14A	0.8801	0.7574	0.1088	0.036*	
H14B	0.9463	0.6140	0.0528	0.036*	
H14C	0.8138	0.5037	0.0720	0.036*	
C10	0.8481 (2)	0.9264 (5)	-0.03388 (12)	0.0242 (7)	
H10	0.9220	0.8353	-0.0446	0.029*	

## supplementary materials

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C11	0.7908 (2)	1.1116 (5)	-0.07566 (12)	0.0264 (7)
H11	0.8257	1.1480	-0.1141	0.032*
C12	0.6822 (2)	1.2436 (5)	-0.06095 (13)	0.0235 (7)
H12	0.6421	1.3704	-0.0894	0.028*
C13	0.6326 (2)	1.1905 (5)	-0.00504 (12)	0.0205 (7)
H13	0.5579	1.2807	0.0049	0.025*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0195 (15)	0.0157 (16)	0.0256 (18)	-0.0019 (12)	0.0003 (13)	0.0007 (13)
O1	0.0434 (12)	0.0128 (11)	0.0326 (12)	0.0013 (8)	0.0142 (9)	0.0032 (8)
C2	0.0229 (15)	0.0104 (13)	0.0197 (16)	-0.0033 (12)	0.0005 (13)	-0.0007 (11)
C7	0.0246 (16)	0.0188 (15)	0.0253 (18)	-0.0012 (13)	0.0015 (14)	-0.0003 (12)
F1	0.0306 (11)	0.0303 (11)	0.0279 (12)	0.0111 (9)	0.0007 (9)	0.0041 (8)
C3	0.0245 (17)	0.0213 (16)	0.0268 (19)	-0.0004 (12)	0.0028 (14)	-0.0024 (12)
F1'	0.050 (8)	0.044 (8)	0.046 (9)	0.018 (6)	0.011 (6)	0.015 (6)
C4	0.0394 (18)	0.0271 (16)	0.0236 (19)	-0.0024 (14)	-0.0015 (15)	0.0038 (13)
C5	0.0409 (18)	0.0276 (16)	0.0170 (17)	-0.0117 (14)	0.0051 (14)	0.0006 (12)
C6	0.0278 (16)	0.0292 (16)	0.0246 (18)	-0.0018 (13)	0.0060 (13)	-0.0021 (13)
N1	0.0258 (13)	0.0101 (13)	0.0239 (15)	0.0023 (11)	0.0060 (11)	0.0022 (11)
C8	0.0203 (15)	0.0113 (14)	0.0206 (17)	-0.0051 (11)	0.0028 (13)	-0.0002 (11)
C9	0.0212 (15)	0.0125 (14)	0.0216 (17)	-0.0031 (11)	-0.0008 (13)	-0.0027 (11)
C14	0.0244 (15)	0.0227 (15)	0.0259 (18)	0.0031 (12)	0.0022 (13)	-0.0016 (12)
C10	0.0234 (15)	0.0241 (15)	0.0255 (18)	0.0002 (13)	0.0040 (13)	-0.0049 (13)
C11	0.0305 (17)	0.0291 (16)	0.0199 (17)	-0.0047 (13)	0.0043 (14)	-0.0017 (13)
C12	0.0295 (16)	0.0188 (15)	0.0216 (18)	-0.0042 (13)	-0.0033 (13)	0.0034 (12)
C13	0.0173 (14)	0.0163 (14)	0.0278 (19)	-0.0020 (11)	-0.0007 (13)	-0.0003 (12)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

C1—O1	1.231 (3)	N1—C8	1.416 (3)
C1—N1	1.347 (3)	N1—HN1	0.85 (3)
C1—C2	1.501 (3)	C8—C13	1.397 (3)
C2—C3	1.386 (3)	C8—C9	1.404 (3)
C2—C7	1.390 (3)	C9—C10	1.385 (4)
C7—F1'	1.292 (10)	C9—C14	1.512 (3)
C7—C6	1.375 (4)	C14—H14A	0.9800
C7—H7	0.9500	C14—H14B	0.9800
F1—C3	1.354 (3)	C14—H14C	0.9800
C3—C4	1.370 (4)	C10—C11	1.384 (3)
C3—H3	0.9500	C10—H10	0.9500
C4—C5	1.385 (4)	C11—C12	1.385 (4)
C4—H4	0.9500	C11—H11	0.9500
C5—C6	1.379 (4)	C12—C13	1.375 (4)
C5—H5	0.9500	C12—H12	0.9500
C6—H6	0.9500	C13—H13	0.9500
O1—C1—N1	123.9 (2)	C13—C8—N1	121.4 (2)

O1—C1—C2	119.7 (2)	C9—C8—N1	118.4 (2)
N1—C1—C2	116.5 (2)	C13—C8—C14 <sup>i</sup>	76.61 (13)
C3—C2—C7	116.7 (2)	C9—C8—C14 <sup>i</sup>	90.97 (14)
C3—C2—C1	124.5 (2)	N1—C8—C14 <sup>i</sup>	103.13 (14)
C7—C2—C1	118.6 (2)	C10—C9—C8	117.7 (2)
F1'—C7—C6	118.8 (6)	C10—C9—C14	121.2 (2)
F1'—C7—C2	119.7 (6)	C8—C9—C14	121.1 (2)
C6—C7—C2	121.4 (2)	C9—C14—H14A	109.5
C6—C7—H7	119.3	C9—C14—H14B	109.5
C2—C7—H7	119.3	H14A—C14—H14B	109.5
F1—C3—C4	117.6 (2)	C9—C14—H14C	109.5
F1—C3—C2	119.1 (2)	H14A—C14—H14C	109.5
C4—C3—C2	123.2 (2)	H14B—C14—H14C	109.5
C4—C3—H3	118.4	C11—C10—C9	122.1 (2)
C2—C3—H3	118.4	C11—C10—H10	118.9
C3—C4—C5	118.4 (3)	C9—C10—H10	118.9
C3—C4—H4	120.8	C10—C11—C12	119.5 (3)
C5—C4—H4	120.8	C10—C11—H11	120.2
C6—C5—C4	120.2 (3)	C12—C11—H11	120.2
C6—C5—H5	119.9	C13—C12—C11	119.8 (2)
C4—C5—H5	119.9	C13—C12—H12	120.1
C5—C6—C7	120.0 (3)	C11—C12—H12	120.1
C5—C6—H6	120.0	C12—C13—C8	120.6 (2)
C7—C6—H6	120.0	C12—C13—C14 <sup>i</sup>	88.08 (15)
C1—N1—C8	125.5 (2)	C8—C13—C14 <sup>i</sup>	81.74 (14)
C1—N1—HN1	115.0 (18)	C12—C13—H13	119.7
C8—N1—HN1	119.2 (18)	C8—C13—H13	119.7
C13—C8—C9	120.2 (2)	C14 <sup>i</sup> —C13—H13	100.3
O1—C1—C2—C3	-140.1 (3)	C1—N1—C8—C13	-37.4 (3)
N1—C1—C2—C3	39.8 (3)	C1—N1—C8—C9	143.2 (2)
O1—C1—C2—C7	36.3 (3)	C1—N1—C8—C14 <sup>i</sup>	44.8 (3)
N1—C1—C2—C7	-143.8 (2)	C13—C8—C9—C10	0.0 (3)
C3—C2—C7—F1'	176.9 (6)	N1—C8—C9—C10	179.4 (2)
C1—C2—C7—F1'	0.3 (7)	C14 <sup>i</sup> —C8—C9—C10	-75.1 (2)
C3—C2—C7—C6	-1.4 (3)	C13—C8—C9—C14	179.9 (2)
C1—C2—C7—C6	-178.0 (2)	N1—C8—C9—C14	-0.6 (3)
C7—C2—C3—F1	-176.9 (2)	C14 <sup>i</sup> —C8—C9—C14	104.9 (2)
C1—C2—C3—F1	-0.5 (3)	C8—C9—C10—C11	0.6 (3)
C7—C2—C3—C4	0.5 (4)	C14—C9—C10—C11	-179.4 (2)
C1—C2—C3—C4	176.9 (2)	C9—C10—C11—C12	-0.7 (4)
F1—C3—C4—C5	178.2 (2)	C10—C11—C12—C13	0.3 (4)
C2—C3—C4—C5	0.7 (4)	C11—C12—C13—C8	0.3 (3)
C3—C4—C5—C6	-1.0 (4)	C11—C12—C13—C14 <sup>i</sup>	79.5 (2)
C4—C5—C6—C7	0.2 (4)	C9—C8—C13—C12	-0.4 (3)
F1'—C7—C6—C5	-177.3 (6)	N1—C8—C13—C12	-179.8 (2)
C2—C7—C6—C5	1.0 (4)	C14 <sup>i</sup> —C8—C13—C12	82.8 (2)

## supplementary materials

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O1—C1—N1—C8	-0.8 (4)	C9—C8—C13—C14 <sup>i</sup>	-83.2 (2)
C2—C1—N1—C8	179.30 (19)	N1—C8—C13—C14 <sup>i</sup>	97.4 (2)

Symmetry codes: (i)  $x, y+1, z$ .

### Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—HN1 $\cdots$ O1 <sup>ii</sup>	0.85 (3)	2.08 (3)	2.887 (3)	158 (2)
C6—H6 $\cdots$ F1 <sup>iii</sup>	0.95	2.60	3.219 (11)	123
C12—H12 $\cdots$ O1 <sup>iv</sup>	0.95	2.63	3.434 (3)	143
C12—H12 $\cdots$ F1 <sup>iv</sup>	0.95	2.40	3.287 (11)	154
C14—H14C $\cdots$ Cg1 <sup>ii</sup>	0.98	2.98	3.702 (3)	131

Symmetry codes: (ii)  $x, y-1, z$ ; (iii)  $-x+1/2, y-1/2, -z+1/2$ ; (iv)  $-x+1, -y+3, -z$ .



Fig. 1

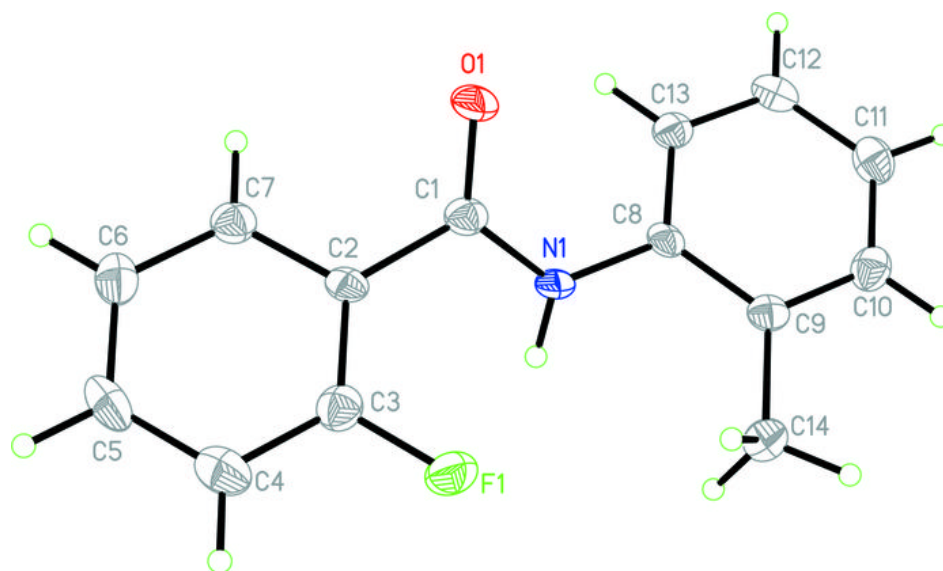


Fig. 2

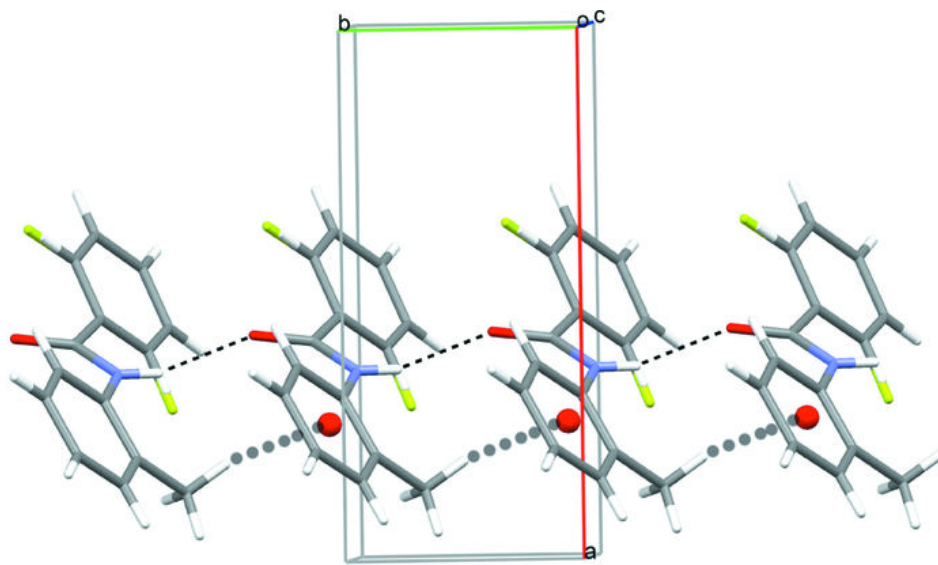


Fig. 3

